## **Practice**

## Logarithmic Functions

Write each equation in exponential form.

$$1.\log_3 81 = 4$$

**2.** 
$$\log_8 2 = \frac{1}{3}$$

3. 
$$\log_{10} \frac{1}{100} = -2$$

Write each equation in logarithmic form.

4. 
$$3^3 = 27$$

5. 
$$5^{-3} = \frac{1}{125}$$

**6.** 
$$\left(\frac{1}{4}\right)^{-4} = 256$$

Evaluate each expression.

7. 
$$\log_7 7^3$$

**10.** 
$$\log_4 32$$

12. 
$$\log_6 \frac{1}{216}$$

Solve each equation.

13. 
$$\log_x 64 = 3$$

**14.** 
$$\log_4 0.25 = x$$

**15.** 
$$\log_4 (2x - 1) = \log_4 16$$

**16.** 
$$\log_{10} \sqrt{10} = x$$

**17.** 
$$\log_7 56 - \log_7 x = \log_7 4$$

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$$\log_7 56 - \log_7 x = \log_7 4$$
 **18.**  $\log_5 (x+4) + \log_5 x = \log_5 12$ 

19. Chemistry How long would it take 100,000 grams of radioactive iodine, which has a half-life of 60 days, to decay to 25,000 grams? Use the formula  $N=N_0\left(\frac{1}{2}\right)^t$ , where N is the final amount of the substance,  $N_0$  is the initial amount, and t represents the number of half-lives.